This article was downloaded by: On: 23 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713617200

Expeditious Synthesis of a Trisubstrate Analogue for

$\alpha(1 \rightarrow 3)$ Fucosyltransferase

B. M. Heskamp^a; G. A. van der Marel^a; J. H. van Boom^a ^a Leiden Institute of Chemistry, Gorlaeus Laboratories, Leiden University, Leiden, RA, The Netherlands

To cite this Article Heskamp, B. M., van der Marel, G. A. and van Boom, J. H.(1995) 'Expeditious Synthesis of a Trisubstrate Analogue for $\alpha(1\rightarrow 3)$ Fucosyltransferase', Journal of Carbohydrate Chemistry, 14: 9, 1265 – 1277 To link to this Article: DOI: 10.1080/07328309508005410 URL: http://dx.doi.org/10.1080/07328309508005410

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

EXPEDITIOUS SYNTHESIS OF A TRISUBSTRATE ANALOGUE

FOR $\alpha(1\rightarrow 3)$ FUCOSYLTRANSFERASE

B.M. Heskamp, G.A.van der Marel and J.H. van Boom*

Leiden Institute of Chemistry, Gorlaeus Laboratories, Leiden University, P.O. Box 9502, 2300 RA, Leiden, The Netherlands

Received March 9, 1995 - Final Form July 21, 1995

ABSTRACT

The synthesis of cyclohexyl 2-acetamido-2-deoxy-3-O-{2-O-[2-(guanosine 5'-O-phosphate)ethyl]- α -L-fucopyranosyl}- β -D-glucopyranoside (1), a potential inhibitor of $\alpha(1\rightarrow 3)$ fucosyltransferases, is described. Target compound 1 was assembled *via* fucosylation of cyclohexyl 2-acetamido-2-deoxy-4,6-O-isopropylidene- β -D-glucopyranoside (6) with ethyl 2-O-[2-(benzoylhydroxy)ethyl]-3,4-O-isopropylidene-1-thio- β -L-fucopyranoside (5) followed by debenzoylation, subsequent condensation of the resulting compound with 3',4'-di-O-benzoyl-5'-O-(2-cyanoethyl-N,N-diisopropylphosphoramidite)-2-N-diphenylacetylguanosine (10) and deprotection.

INTRODUCTION

Recently it has been revealed that the sialyl Lewis X (SLe^X: Neu5Ac $(\alpha 2\rightarrow 3)$ Gal($\beta 1\rightarrow 4$)[Fuc($\alpha 1\rightarrow 3$)]GlcNAc) determinant and other naturally occurring derivatives thereof are involved in the interaction with selectins (*i.e.* E-selectin, L-selectin and P-selectin).¹ The recognition of SLe^X located on leucocyte surfaces by E-selectin, which is expressed on activated endothelial cells, is a critical event in the adhesion of leucocytes to the vascular endothelium. The latter process eventually results in the migration of leucocytes from the bloodstream to sites of inflammation.

Since excessive leucocyte recruitment is related to inflammatory diseases (*e.g.* rheumatoid arthritis), obstruction of the E-selectin-SLe^x interaction may be a promising approach towards the development of anti-inflammatory drugs. Biological studies² showed that the fucose residue is a crucial structural element of SLe^x for the binding with E-selectin. Incorporation of fucose is the final step in the biosynthesis of SLe^x and is mediated by a specific $\alpha(1\rightarrow3)$ fucosyltransferase³ [$\alpha(1\rightarrow3)$ FucT]. Consequently, with the aim to develop anti-inflammatory therapeutics, effort has been devoted to the preparation of $\alpha(1\rightarrow3)$ FucT inhibitors.⁴ In order to achieve this goal, metabolically stable *substrate analogues* of guanosine 5' -(β -L-fucopyranosyl)-diphosphate (GDP-Fuc), the naturally occurring donor substrate of fucosyltransferases, have been prepared.^{4b-d} Apart from this, Hindsgaul *et al.*⁵ reported a second type of inhibitor for $\alpha(1\rightarrow2)$ FucT, a so-called *bisubstrate analogue*. This analogue, which is characterized by the presence of a nucleotide (GDP) as well as an acceptor (phenyl β -D-galactopyranoside) residue, showed a slightly increased inhibitory effect on $\alpha(1\rightarrow2)$ FucT compared to the endogenous inhibitor GDP.

As part of a program directed towards the development of $\alpha(1\rightarrow 3)$ FucT inhibitors, we here describe the design and preparation of *trisubstrate analogue* 1.



RESULTS AND DISCUSSION

It has been postulated⁵ that the enzymatic transfer of a fucosyl moiety to an acceptor proceeds *via* an ion-pair mechanism, implying that the donor (GDP-Fuc) and acceptor are interacting simultaneously with the enzyme in the transition state. On the basis of this hypothesis we reasoned that incorporation of three important recognition sites (*i.e.* a guanosine, a fucose and an acceptor moiety) in a molecule may result in an active and selective inhibitor. One possible route to validate this

concept entails the design and synthesis of *trisubstrate analogue* 1, in which a L-fucopyranosyl residue is α -linked to the HO-3 function of cyclohexyl 2-acetamido-2-deoxy- β -D-glucopyranoside (*i.e.* the acceptor) while a guanosine moiety is anchored to the HO-2 group of the fucose *via* a phosphate-ethylene bridge. The flexibility of the spacer may enable the guanosine and fucose residues to adopt the correct relative orientation for optimal recognition by the enzyme.

The synthetic route to target compound 1 is outlined in Scheme 1 and comprises four stages: (i) preparation of the hydroxyethyl spacer containing fucosyl donor 5, (ii) coupling of donor 5 to glucosamine acceptor 6, (iii) introduction of the guanosine 5'-phosphate residue and finally (iv) cleavage of protective groups.

The requisite fucosyl donor **5** was prepared from readily accessible ethyl 3,4-*O*-isopropylidene-1-thio- β -L-fucopyranoside⁶ **2** by the following three-step procedure (Scheme 1). The 2-*O*-(methoxycarbonylmethyl) substituent, as in **3**, was introduced by treatment of **2** with excess methyl bromoacetate and sodium hydride.⁷ Reduction of the ester function in **3** with lithium aluminium hydride and subsequent benzoylation of the generated hydroxyl function (in **4**) with benzoyl chloride in pyridine afforded ethyl 2-*O*-[2-(benzoylhydroxy)ethyl]-3,4-*O*-isopropylidene-1-thio- β -L-fucopyranoside (**5**) in 66% overall yield.

Having donor 5 in hand, we turned our attention to the glycosylation of acceptor cyclohexyl 2-acetamido-2-deoxy-4.6-O-isopropylidene- β -D-glucopyranoside (6), which was prepared as follows. First, glycosylation of cyclohexanol was achieved with readily accessible 2-methyl- $(3,4,6-tri-O-acetyl-1,2-dideoxy-\alpha-D-acetyl-1,2-dideoxy-acetyl-1,2-di$ glucopyrano)-[2,1-d]-2-oxazoline⁸ under the promotion of catalytic trifluoromethanesulfonic acid (TfOH). Subsequent Zemplén deacetylation and treatment with 2,2-dimethoxypropane and catalytic p-toluenesulfonic acid furnished acceptor 6. Condensation of fucopyranosyl donor 5 with acceptor 6 under the agency⁹ of N-iodosuccinimide (NIS) and catalytic TfOH gave disaccharide 7 as a mixture of anomers ($\alpha/\beta = 4/1$) in 71% yield. On the other hand, a higher yield (94%) and a more favourable α/β ratio (7/1) was obtained by the use of the promoter system cupric bromide and tetrabutylammonium bromide as proposed by Ogawa.¹⁰ Separation of the individual anomers by silica gel chromatography led to the α -linked



Reagents and conditions:

(*i*) NaH, BrCH₂COOMe, imidazole, Bu₄NI, THF, reflux (73%); (*ii*) LiAlH₄, THF, reflux; (*iii*) BzCl, pyridine
(91%, 2 steps); (*iv*) 6, CuBr₂, Bu₄NBr, DCE, DMF (94%, α/β=7/1); (*v*) KOt-Bu, MeOH (91%);
(*vi*) CIP(OCH₂CH₂CN)[N(*i*-Pr)₂], DIPEA, CH₂Cl₂ (59%); (*vii*) (a) 10, 1*i*-tetrazole, CH₂Cl₂, CH₃CN;
(b) *t*-BuOOH; (c) TEA (79%, 3 steps); (*viii*) (a) PPTS, MeOH, 50 °C; (b) NH₄OH, 50 °C (90%, 2 steps).



dimer 7 in 82% yield. Removal of the benzoyl group in 7 by Zemplén deacylation afforded the requisite disaccharide 8.

The next stage in the assembly of target molecule 1 entailed the introduction of the 5'-phosphate guanosine moiety in disaccharide 8. To this end, the primary hydroxyl group of guanosine derivative 9^{11} was phosphitylated with chloro 2cyanoethyl-N,N-diisopropylphosphoramidite¹² in the presence of N.N-3',4'-di-O-benzoyl-5'-O-(2-cyanoethyl-N,Ndiisopropylethylamine furnish to diisopropylphosphoramidite)-2-N-diphenylacetylguanosine (10). Phosphodiester 11 was obtained via a one-pot three-step procedure. Thus, condensation of disaccharide 8 with reagent 10 under the agency of 1H-tetrazole¹³ was followed by in situ oxidation with tert-butyl hydroperoxide. Subsequent elimination of the cyanoethyl

group from the intermediate phosphotriester with triethylamine gave phosphodiester 11 (triethylammonium salt, δ_p -0.54 ppm) in 79% yield.

Finally, the protective groups in trisubstrate analogue 11 were unleashed by the following two-step protocol. Deacetonisation of 11 was effected by treatment with pyridinium *p*-toluene sulfonate (PPTS) in methanol. Saponification of the ester functions with aqueous ammonia and subsequent purification by gel filtration furnished target compound 1 in 90% yield. The structure of 1 was firmly established by ¹H 2D and CH COSY NMR experiments.

In conclusion, a straightforward synthesis of *trisubstrate analogue* 1 has been described. The synthetic route can readily be adopted for the assembly of *trisubstrate analogues* containing alternative spacers (*e.g.* a neutral phosphate bridge¹⁴ or an isosteric spacer¹⁵) or different acceptor moieties in order to further assess the biological activity of this type of *trisubstrate analogues*. The inhibitory effect of compound 1 on $\alpha(1\rightarrow3)$ FucT-VI is currently under investigation and will be reported in due course.

EXPERIMENTAL

General procedures. 1,2-Dichloroethane, dichloromethane, diethyl ether and toluene were distilled from P_2O_5 (5 g/L). Acetonitrile and pyridine were dried by refluxing overnight with CaH₂ (5 g/L) and then distilled. *N,N*-Dimethylformamide (DMF) was stirred overnight with CaH₂ (5 g/L) and then distilled under reduced pressure. Methanol was dried by refluxing with magnesium methoxide and then distilled. Tetrahydrofuran (THF) was distilled from LiAlH₄. Methanol was stored on 0.3 nm and all other solvents on 0.4 nm molecular sieves. Schleicher and Schüll DC Fertigfolien F 1500 LS 254 were used for TLC-analysis. Compounds were visualized by UV light (254 nm) and by charring with 20% sulfuric acid in methanol. Column chromatography was performed on silica gel 60, 230-400 mesh (Merck). The petroleum ether used for elution during chromatography was light boiling (40-60 °C). Gel filtration was performed on Fractogel TSK HW-40. ¹H NMR (200 MHz), ¹³C NMR (50.1 MHz) and ³¹P NMR (80.7 MHz) spectra were recorded using a Jeol JNM-FX 200 spectrometer, unless stated otherwise. ¹H NMR (300 MHz) spectra

were recorded using a Bruker WM-300 spectrometer, ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded using a Bruker MSL-400 spectrometer. ¹H and ¹³C chemical shifts (δ) are given in ppm relative to tetramethylsilane as internal standard, ³¹P chemical shifts are given in ppm relative to 85% H₃PO₄ as external standard.

Ethyl 3,4-O-Isopropylidene-2-O-(methoxycarbonylmethyl)-1-thio-β-Lfucopyranoside (3). NaH (600 mg 60% dispersion, 15 mmol), imidazole (20 mg) and Bu_4NBr (20 mg) were added to a solution of compound 2 (1.38 g, 5.56 mmol), which was dried previously by evaporation with toluene (3×10 mL), in THF (25 mL). After stirring for 30 min, methyl bromoacetate (1.00 mL, 10.56 mmol) was added and the mixture was heated at reflux. After 16 h TLC analysis indicated complete conversion of the starting material to a less polar compound and the cooled mixture was quenched with methanol. The mixture was neutralized with dilute aqueous HCl and extracted with dichloromethane. The organic layer was washed with saturated aqueous NaHCO₃ and brine, dried (MgSO₄) and concentrated in vacuo. The crude product was purified by column chromatography (diethyl ether/petroleum ether, 0:1 to 3:7 v/v) to give 1.30 g (73%) of pure 3. ^{1}H NMR (CDCl₂) δ 1.30 (t, 3H, SCH_2CH_3 , 1.34, 1.50 (2s, 6H, $C(CH_3)_2$), 1.39 (d, 3H, H-6, $J_{5,6} = 6.6$ Hz), 2.74 (m, 2H, SCH₂CH₃), 3.45 (dd, 1H, H-2, $J_{1,2} = 9.6$ Hz, $J_{2,3} = 6.9$ Hz), 3.76 (s, 3H, OCH₃), 3.81 (dq, 1H, H-5, $J_{4.5} = 2.1$ Hz), 4.04 (dd, 1H, H-4, $J_{3.4} = 5.6$ Hz), 4.21 (dd, 1H, H-3), 4.39 (d, 1H, H-1), 4.42 (AB, 2H, H-7); ${}^{13}C{}^{1}H{}$ NMR (CDCl₃) δ 13.7 (SCH₂CH₃), 14.3 (C-6), 23.8 (SCH₂CH₃), 25.8, 27.5 (C(CH₃)₂), 51.2 (OCH₃), 67.8 (C-7), 71.9, 76.0, 78.7, 79.7 (C-2, C-3, C-4, C-5), 82.3 (C-1), 109.1 (C(CH₃)₂), 169.8 (C=O).

Ethyl 2-O-(2-Hydroxyethyl)-3,4-O-isopropylidene-1-thio- β -L-fucopyranoside (4). Compound 3 (1.30 g, 4.06 mmol) was dried by evaporation with toluene (3 \times 10 mL) and dissolved in THF (20 mL). LiAlH₄ (600 mg, 15.79 mmol) was added and the mixture was heated at reflux. After 16 h TLC analysis showed completion of the reaction and excess LiAlH₄ was decomposed with ethyl acetate (45 mL). A 20% KOH solution (60 mL) and water (100 mL) were added and the mixture was extracted with diethyl ether (2 \times 30 mL). The combined organic layers were washed with brine, dried (MgSO₄) and concentrated *in vacuo* to give 4 in a quantitative yield (1.19 g). ¹H NMR (CDCl₃) δ 1.30 (t, 3H, SCH₂CH₃), 1.36, 1.54 (2s, 6H, C(CH₃)₂), 1.40 (d, 3H, H-6, J_{5,6} = 6.6 Hz), 2.74 (m, 2H, SCH₂CH₃), 2.95 (bs, 1H, OH), 3.33 (dd, 1H, H-2, J_{1,2} = 10.1 Hz, J_{2,3} = 6.6 Hz), 3.71-3.88 (m, 5H, H-3, H-4, H-5, H-8), 4.04-4.21 (m, 2H, H-7), 4.28 (d, 1H, H-1); ¹³C{¹H} NMR (CDCl₃) δ 14.1 (SCH₂CH₃), 15.9 (C-6), 23.4 (SCH₂CH₃), 25.4, 27.2 (C(CH₃)₂), 60.7 (C-8), 72.9 (C-7), 71.5, 75.6, 78.6, 79.7 (C-2, C-3, C-4, C-5), 82.5 (C-1), 108.7 (C(CH₃)₂).

Ethyl 2-*O*-[2-(Benzoyloxy)ethyl]-3,4-*O*-isopropylidene-1-thio-β-Lfucopyranoside (5). Compound 4 (4.06 mmol) was dried by evaporation with pyridine (3 × 10 mL), dissolved in pyridine (20 mL) and benzoylchloride (0.50 mL, 4.31 mmol) was added. After stirring for 1 h the mixture was quenched with water and concentrated *in vacuo*. The residue was redissolved in dichloromethane, the solution was washed with saturated aqueous NaHCO₃, water, dried (MgSO₄) and concentrated *in vacuo*. The crude product was purified by column chromatography (diethyl ether/petroleum ether, 0:1 to 3:7 v/v) to give 1.61 g (92%) of pure 5. R_f 0.87 (diethyl ether); ¹H NMR (CDCl₃) δ 1.25 (t, 3H, SCH₂CH₃), 1.34, 1.53 (2s, 6H, $C(CH_3)_2$), 1.39 (d, 3H, H-6, J_{5,6} = 6.6 Hz), 2.70 (q, 2H, SCH₂CH₃), 3.36 (dd, 1H, H-2, J_{1,2} = 9.9 Hz, J_{2,3} = 6.6 Hz), 3.80 (dq, 1H, H-5, J_{4,5} = 2.0 Hz), 4.02-4.19 (m, 4H, H-3, H-4, H-7), 4.34 (d, 1H, H-1), 4.48 (t, 2H, H-8), 7.38-8.13 (m, 5H, H_{arom}.); ¹³C{¹H} NMR (CDCl₃) δ 14.1 (SCH₂CH₃), 16.1 (C-6), 23.6 (SCH₂CH₃), 25.6, 27.4 (C(CH₃)₂), 63.3 (C-8), 69.2 (C-7), 71.6, 75.7, 78.8, 80.2 (C-2, C-3, C-4, C-5), 82.4 (C-1), 108.7 (*C*(CH₃)₂), 127.5, 129.0, 132.1 (CH_{arom}), 129.6 (C_{arom}), 165.5 (C=O).

Anal. Calcd for $C_{20}H_{28}O_6S$ (396.51): C, 60.59; H, 7.12. Found: C, 60.73; H, 7.19.

Cyclohexyl 2-Acetamido-2-deoxy-4,6-*O*-isopropylidene-β-D-glucopyranoside (6). 2-Methyl-(3,4,6-tri-*O*-acetyl-1,2-dideoxy-α-D-glucopyrano)-[2,1-*d*]-2oxazoline (7.9 g, 24.0 mmol) was dried by evaporation with toluene and dissolved in dichloromethane (50 mL). Cyclohexanol (3.7 mL, 35.6 mmol) and TfOH (210 μ L, 2.4 mmol) were added and the mixture was stirred for 2 h, when TLC analysis indicated completion of the reaction. The mixture was neutralized with TEA, washed with saturated aqueous NaHCO₃ and water, dried (MgSO₄) and concentrated *in vacuo*. The crude crystalline product was recrystallized from ethanol to give pure cyclohexyl 2-acetamido-3,4,6-tri-*O*-acetyl-2-deoxy-β-D-glucopyranoside (6.9 g, 67%).

The peracetylated glycoside (6.9 g, 16.1 mmol) was dissolved in methanol (50 mL) and NaOMe (100 mg) was added. After 2 h the mixture was neutralized with Dowex 50 W X 4 [H⁺], filtered and the filtrate was concentrated in vacuo. The residual methanol was removed by evaporation with toluene $(3 \times 10 \text{ mL})$. The deacetylated product was dissolved in acetone (50 mL) and dimethoxypropane (15 mL) and the acidity of the solution was adjusted to pH 4 with p-TsOH. After stirring for 4 h at 40 °C TLC analysis indicated disappearance of the starting material and the mixture was neutralized with TEA. The solvent was evaporated, the residue was redissolved in dichloromethane and washed with saturated aqueous NaHCO3 and water, dried (MgSO₄) and concentrated in vacuo. The crude product was purified by column chromatography (ethanol/petroleum ether, 0:1 to 5:95 v/v) to give pure 6. Yield: 4.6 g (83%, 2 steps); $R_f 0.54$ (methanol/dichloromethane, 1:9 v/v); ¹H NMR $(CDCl_3)$ δ 1.17-1.91 (m, 10H, 5 × CH₂ Hex), 1.44, 1.52 (2s, 6H, C(CH₃)₂), 2.04 (s, 3H, COCH₃), 3.03 (dt, 1H, H-5, J = 9.9 Hz, $J_{5.6b}$ = 5.6 Hz), 3.32 (dd, 1H, H-2, $J_{NH,2}$ = 5.35 Hz, $J_{2.3}$ = 9.6 Hz), 3.56-3.68 (m, 1H, OCH cHex), 3.58 (t, 1H, H-4, J = 9.3 Hz), 3.80 (t, 1H, H-6a, J = 10.3 Hz), 3.92 (dd, 1H, H-6b, $J_{6a,6b} = -10.3$ Hz), 4.04 (t, 1H, H-3), 4.77 (d, 1H, H-1, $J_{1,2} = 8.1$ Hz), 5.77 (bs, 1H, NH); ${}^{13}C{}^{1}H$ NMR $(CDCl_3)$ δ 18.7, 28.6 $(C(CH_3)_2)$, 22.9 $(COCH_3)$, 23.4, 23.6, 25.1, 31.4, 33.0 (5×10^{-6}) CH₂ cHex), 57.6 (C-2), 61.7 (C-6), 66.6, 70.8, 73.9, 77.3 (C-3, C-4, C-5, OCH cHex), 99.2 (C(CH₃)₂), 99.2 (C-1), 174.9 (C=O).

Anal. Calcd for C₁₇H₂₉O₆N (343.42): C, 59.46; H, 8.51; N, 4.08. Found: C, 59.54; H, 8.47; N, 4.15.

Cyclohexyl 2-Acetamido-{3-O-[2-(benzoyloxy)ethyl]-3,4-O-isopropylidene- α -L-fucopyranosyl}-2-deoxy-4,6-O-isopropylidene- β -D-glucopyranoside (7). Donor 5 (500 mg, 1.26 mmol) and acceptor 6 (500 mg, 1.46 mmol) were dried by evaporation with 1,2-dichloroethane (3 × 5 mL), dissolved in 1,2-dichloroethane (5 mL) and DMF (1 mL) and stirred for 30 min with 0.4 nm molecular sieves under a nitrogen atmosphere. CuBr₂ (420 mg, 1.88 mmol) and Bu₄NBr (600 mg, 1.86 mmol) were added and stirring was continued. After 16 h TLC analysis showed complete disappearance of the donor and the reaction mixture was diluted with dichloromethane and filtered. The filtrate was washed with saturated aqueous NaHCO₃, brine, dried (MgSO₄) and concentrated *in vacuo*. Purification by column Downloaded At: 08:48 23 January 2011

chromatography (diethyl ether/petroleum ether, 1:1 to 1:0 v/v) gave 700 mg of pure α -linked dimer 7 and 105 mg of the β -anomer (total yield: 94%; α/β 7/1). R_f (α) 0.71 (methanol/dichloromethane, 1:9 v/v); ¹H NMR (300 MHz 2D COSY) (CDCl₃) δ 1.18-1.89 (m, 28H, Fuc: H-6, $2 \times C(CH_3)_2$, COCH₃, $5 \times CH_2$ cHex), 2.95 (ddd, 1H, GlcNAc: H-2, $J_{1,2} = 8.3$ Hz, $J_{2,NH} = 7.2$ Hz, $J_{2,3} = 9.9$ Hz), 3.25 (dt, 1H, GlcNAc: H-5, $J_{4.5} = 9.9$ Hz, $J_{5.6b} = 5.4$ Hz), 3.44 (t, 1H, GlcNAc: H-4, J = 9.8 Hz), 3.46 (dd, 1H, Fuc: H-2, $J_{1,2} = 3.6$ Hz, $J_{2,3} = 8.0$ Hz), 3.55-3.62 (m, 1H, OCH cHex), 3.73 (t, 1H, GlcNAc: H-6a, J = 10.5 Hz), 3.83 (ddd, 1H, Fuc: H-7a, $J_{7a,8a} = 3.1$ Hz, $J_{7a,8b} =$ 5.7 Hz, $J_{7a,7b} = -11.1$ Hz), 3.87 (t, 1H, GlcNAc: H-6b), 3.99 (dd, 1H, Fuc: H-4, $J_{3,4}$ = 5.5 Hz, J = 2.6 Hz), 4.17 (t, 1H, GlcNAc: H-3), 4.19 (dd, 1H, Fuc: H-3), 4.22 (ddd, 1H, Fuc: H-7b, $J_{7b,8a} = 7.1$ Hz, $J_{7b,8b} = 3.0$ Hz), 4.41-4.49 (m, 2H, Fuc: H-5, H-8a), 4.73 (ddd, 1H, Fuc: H-8b, $J_{8a,8b} = -11.9$ Hz), 4.83 (d, 1H, Fuc: H-1), 5.16 (d, 1H, GlcNAc: H-1), 6.52 (d, 1H, NH), 7.45-8.07 (m, 5H, $H_{arom.}$); ¹³C{¹H} NMR (CDCl_3) δ 16.1 (Fuc: C-6), 19.1, 26.3, 28.3, 29.0 (2 × C(CH_3)_2), 23.3 (COCH_3), 23.6, 23.8, 25.5, 31.7, 33.3 (5 × CH_2 cHex), 59.2 (GlcNAc: C-2), 62.2 (GlcNAc: C-6), 62.7, 66.5, 73.3, 76.2, 77.4, 78.8 (Fuc: C-2, C-3, C-4, C-5, GlcNAc: C-3, C-4, C-5, OCH cHex), 64.0 (Fuc: C-8), 69.0 (Fuc: C-7), 98.2 (Fuc, GlcNAc: C-1), 98.9, 108.7 (2 × $C(CH_3)_2$), 128.5, 129.6, 133.2 (CH_{arom}), 170.6 (C=O).

Anal. Calcd for $C_{35}H_{51}O_{12}N$ (677.80): C, 62.02; H, 7.58; N, 2.07. Found: C, 62.19; H, 7.66; N, 1.99.

7 β : ¹³C{¹H} NMR (CDCl₃) δ 16.5 (Fuc: C-6), 18.8, 26.0, 27.9, 28.9 (2 × C(CH₃)₂), 23.2 (COCH₃), 23.6, 23.8, 25.4, 31.6, 33.3 (5 × CH₂ cHex), 57.7 (GlcNAc: C-2), 62.2 (GlcNAc: C-6), 64.2, 69.4 (Fuc: C-7, C-8), 66.5, 68.2, 74.3, 75.5, 76.1, 77.3, 81.3 (Fuc: C-2, C-3, C-4, C-5, GlcNAc: C-3, C-4, C-5, OCH cHex), 98.8 (GlcNAc: C-1), 99.1, 109.3 (2 × C(CH₃)₂), 102.3 (Fuc: C-1), 128.1, 129.5, 132.7 (CH_{arom}), 170.6 (C=O).

Cyclohexyl 2-Acetamido-2-deoxy-3-O-[2-O-(2-hydroxyethyl)-3,4-Oisopropylidene- α -L-fucopyranosyl]-4,6-O-isopropylidene- β -D-glucopyranoside (8). Disaccharide 7 (155 mg, 0.23 mmol) was dissolved in methanol (2 mL) and KOt-Bu (10 mg, 0.09 mmol) was added. After 16 h the mixture was neutralized with Dowex 50 W X 4 [H⁺], filtered and concentrated *in vacuo* to dryness. The crude product was purified by column chromatography (methanol/dichloromethane, 0:1 to 5:95 v/v) to give 120 mg (91%) of pure **8**. $R_f 0.50$ (methanol/dichloromethane, 1:9 v/v); ¹H NMR (CDCl₃) δ 1.21-1.90 (m, 10H, 5 × CH₂ cHex), 1.27 (d, 3H, Fuc: H-6, $J_{5,6} = 6.6$ Hz), 1.33, 1.39, 1.43, 1.50 (4s, 12H, 2 × C(CH₃)₂), 1.96 (s, 3H, COCH₃), 3.16 (bq, 1H, GlcNAc: H-2, J = 7.7 Hz), 3.39 (dt, 1H, GlcNAc: H-5, J = 8.8 Hz, $J_{5,6b} = 4.4$ Hz), 3.50-3.94 (m, 9H, Fuc: H-2, H-7, H-8, GlcNAc: H-4, H-6, OCH cHex), 4.00 (dd, 1H, Fuc: H-4, $J_{3,4} = 5.4$ Hz, $J_{4,5} = 2.6$ Hz), 4.19 (dd, 1H, Fuc: H-3, $J_{2,3} = 8.2$ Hz), 4.28 (t, 1H, GlcNAc: H-3, J = 9.4 Hz), 4.53 (dq, 1H, Fuc: H-5), 4.98 (d, 1H, Fuc: H-1, $J_{1,2} = 3.6$ Hz), 5.13 (d, 1H, GlcNAc: H-1, $J_{1,2} = 8.4$ Hz), 6.58 (d, 1H, NH); 13C{1H} NMR (CDCl₃) δ 16.1 (Fuc: C-6), 19.1, 26.2, 28.2, 28.8 (2 × C(CH₃)₂), 23.3 (COCH₃), 23.8, 25.3, 31.8, 33.3 (5 × CH₂ cHex), 58.5 (GlcNAc: C-2), 61.7 (Fuc: C-8), 62.2 (GlcNAc: C-6), 62.5, 66.7, 73.0, 75.4, 76.1, 77.5, 78.1 (Fuc: C-2, C-3, C-4, C-5, GlcNAc: C-3, C-4, C-5, OCH cHex), 71.8 (Fuc: C-7), 96.9 (Fuc: C-1), 98.3 (GlcNAc: C-1), 99.0, 108.5 (2 × C(CH₃)₂), 171.1 (C=O).

Anal. Calcd for $C_{28}H_{47}O_{11}N$ (573.69): C, 58.62; H, 8.26; N, 2.44. Found: C, 58.69; H, 8.37; N, 2.42.

3',4'-Di-O-benzoyl-5'-O-(2-cyanoethyl-N,N-diisopropylphosphoramidite)-2-N-diphenylacetylguanosine (10). Compound 9 (1.38 g, 2.00 mmol) was dried by evaporation with 1,2-dichloroethane $(3 \times 5 \text{ mL})$, dissolved in 1,2-dichloroethane (10 mL) and kept under an atmosphere of nitrogen. DIPEA (0.70 mL, 4.02 mmol) and chloro 2-cyanoethyl-N,N-diisopropylphosphoramidite (0.63 mL, 2.80 mmol) were mixture was stirred for 30 min, when TLC analysis added and the (methanol/dichloromethane, 1:9 v/v) indicated complete conversion of the starting material. Water (10 mL) was added and the mixture was extracted with dichloromethane. The organic layer was washed with brine $(2 \times)$ and saturated aqueous NaHCO₃, dried (MgSO₄), and concentrated in vacuo to dryness. The crude product was purified by flash column chromatography (TEA/ethanol/diethyl ether, 1:0:99 to 1:4:95 v/v) to give 1.05 g (59%) of pure 10. $R_{\rm f}$ 0.60 (methanol/dichloromethane, 1:9 v/v); ${}^{31}P{}^{1}H$ NMR (CDCl₃) δ 149.15, 149.82.

Cyclohexyl2-Acetamido-3-O-{2-O-[2-(3',4'-di-O-benzoyl-2-N-diphenylacetylguanosine5'-O-phosphate)ethyl]-3,4-O-isopropylidene- α -L-fucopyranosyl}-2-deoxy-4,6-O-isopropylidene- β -D-glucopyranoside(11).Compound8(120 mg, 0.21 mmol) and 10(230 mg, 0.26 mmol) were dried by

evaporation with toluene $(3 \times 5 \text{ mL})$ and dissolved in dichloromethane (3 mL). 1Htetrazole (60 mg, 0.86 mmol), dried by evaporation with toluene $(3 \times 5 \text{ mL})$ and dissolved in acetonitrile (1 mL), was added. TLC analysis (R_f 0.55, methanol/dichloromethane, 1:9 v/v) indicated completion of the reaction after 10 min and t-BuOOH (115 µL, 0.50 mmol) was added. TLC analysis indicated completion of the oxidation after 30 min and TEA (1 mL) was added. After stirring for 4 h the mixture was concentrated in vacuo and the residue was purified by column chromatography (methanol/dichloromethane, 0:9 to 1:9 v/v) to give 230 mg (77%) of pure 11 in the triethylammonium form. $R_f 0.37$ (methanol/dichloromethane, 1:9 v/v); $^{13}C{^{1}H}$ NMR (MeOD/CDCl₃) δ 8.2 (N(CH₂CH₃)₃), 15.8 (Fuc: C-6), 18.7, 25.8, 27.8, 28.5 (2 × C(CH₃)₂), 22.6 (COCH₃), 23.4, 23.5, 25.0, 31.3, 33.0 (5 × CH₂ cHex), 45.8 (N(CH₂CH₃)₃), 56.7 (CH(Ph)₂), 57.2 (GlcNAc: C-2), 61.8 (GlcNAc: C-6), 62.5, 66.5, 71.8, 72.8, 73.0, 75.2, 75.9, 77.6, 78.1 (Fuc: C-2, C-3, C-4, C-5, GlcNAc: C-3, C-4, C-5, Rib: C-2, C-3, OCH cHex), 64.2, 64.7 (Fuc: C-8, Rib: C-5), 70.3 (Fuc: C-7), 82.4 (d, Rib: C-4, J_{4P} = 8.8 Hz), 86.8 (Rib: C-1), 97.2 (Fuc: C-1), 99.1 (GlcNAc: C-1), 98.9, 108.2 (2 × $C(CH_3)_2$), 120.9 (G: C-5), 124.1-129.2 (CH_{arom}), 137.4, 137.7 (C_{arom}), 133.4 (G: C-8), 147.8, 148.5 (G: C-2, C-4), 155.6 (G: C-6), 164.7, 165.0, 172.1, 175.0 (4 × C=O); ${}^{31}P{}^{1}H{}$ NMR (CH₂Cl₂) δ -0.54.

Anal. Calcd for $C_{72}H_{92}O_{21}N_7P$ (1422.54): C, 60.79; H, 6.52; N, 5.89. Found: C, 60.98; H, 6.61; N, 6.80.

Cyclohexyl 2-Acetamido-2-deoxy-3-O-{2-O-[2-(guanosine 5'-Ophosphate)ethyl]-α-L-fucopyranosyl}-β-D-glucopyranoside (1). Compound 11 (140 mg, 0.098 mmol) was dissolved in methanol (1 mL) and PPTS (10 mg, 0.040 mmol) was added. After 16 h at 50 °C TLC analysis showed complete conversion to a more polar compound, the mixture was concentrated *in vacuo* and redissolved in 25% aqueous NH₄OH. After 48 h at 50 °C the mixture was concentrated *in vacuo*, taken up in water, washed with dichloromethane (3 ×) and lyophilized. The crude product was purified by gel filtration (water/methanol/2 M TEAB, 85:10:5 v/v, R_t 134 min), concentrated *in vacuo* and applied to a column of Dowex 50 W X 4 [Na⁺], which was eluted with water. The resulting solution was concentrated *in vacuo*, the residue was redissolved in D₂O and lyophilized to give 76 mg (90%) of the title compound. R_f 0.41 (*i*-PrOH/NH₄OH/water, 6:3:1 v/v); ¹H NMR (400 MHz 2D COSY, CH COSY) (D₂O) δ 1.18 (d, 3H, Fuc: H-6, J_{5.6} = 6.6 Hz), 1.21-1.94 (m, 10H, 5 × CH₂ cHex), 2.03 (s, 3H, COCH₃), 3.44-3.54 (m, 2H, GlcNAc: H-4, H-5), 3.63 (dd, 1H, Fuc: H-2, $J_{1,2} = 3.9$ Hz, $J_{2,3} = 10.3$ Hz), 3.70-3.76 (m, 6H, Fuc: H-7, GlcNAc: H-2, H-3, H-6a, OCH cHex), 3.80 (bd, 1H, Fuc: H-4, $J_{3,4} = 3.4$ Hz), 3.89 (dd, 1H, Fuc: H-3), 3.92-4.00 (m, 3H, Fuc: H-8, GlcNAc: H-6b), 4.15-4.17 (m, 2H, Rib: H-5), 4.32-4.37 (m, 2H, Fuc: H-5, Rib: H-4), 4.53 (t, 1H, Rib: H-3, J = 4.5 Hz), 4.64 (d, 1H, GlcNAc: H-1, $J_{1,2} = 8.0$ Hz), 4.80 (t, 1H, Rib: H-2), 5.18 (d, 1H, Fuc: H-1), 5.95 (d, 1H, Rib: H-1, $J_{1,2} = 5.7$ Hz), 8.11 (s, 1H, G: H-8); ${}^{13}C{}^{1}H$ NMR (100 MHz) (D₂O) δ 15.9 (Fuc: C-6), 23.3 (COCH₃), 23.7, 23.9, 25.7, 31.7, 33.3 (5 × CH₂) cHex), 56.3 (GlcNAc: C-2), 61.6 (GlcNAc: C-6), 65.7 (d, Rib: C-5, J_{5, P} = 4.4 Hz), 65.9 (d, Fuc: C-8, $J_{8,P}$ = 5.9 Hz), 67.6 (Fuc: C-5), 69.3 (Fuc: C-3), 69.6, 76.4 (GlcNAc: C-4, C-5), 70.5 (d, Fuc: C-7, $J_{7,P}$ = 7.3 Hz), 71.1 (Rib: C-3), 72.5 (Fuc: C-4), 74.4 (Rib: C-2), 77.3 (Fuc: C-3), 79.0, 79.8 (GlcNAc: C-3, OCH cHex), 84.4 (d, Rib: C-4, $J_{4,P} = 9.0$ Hz), 87.6 (Rib: C-1), 97.5 (Fuc: C-1), 100.0 (GlcNAc: C-1), 117.0 (G: C-5), 138.1 (G: C-8), 152.6 (G: C-4), 154.8 (G: C-2), 159.8 (G: C-6), 174.9 (C=O); ${}^{31}P{}^{1}H{}$ NMR (D₂O) δ 0.91.

Anal. Calcd for $C_{32}H_{50}O_{18}N_6NaP$ (860.75): C, 44.65; H, 5.86; N, 9.76. Found: C, 44.57; H, 5.90; N, 9.87.

REFERENCES AND NOTES

- Recent reviews on SLe^x and the selectins: (a) A. Giannis, Angew. Chem. Int. Ed. Engl., 33, 178 (1994); (b) D. E. Levy, P. C. Tang and J. H. Musser, Ann. Rep. Med. Chem., 215 (1994).
- (a) J. Y. Ramphal, Z.-L. Zheng, C. Perez, L. E. Walker, S. A. DeFrees and F. C. A. Gaeta, J. Med. Chem., 37, 3459 (1994); (b) M. Tiemeyer, S. J. Swiedler, M. Ishihara, M. Moreland, H. Schweingruber, P. Hirtzer and B. K. Brandley, Proc. Natl. Acad. Sci. U.S.A. 88, 1138 (1991); (c) J. B. Lowe, L. M. Stoolman, R. P. Nair, R. D. Larsen, T. L. Berhend and R. M. Marks, Cell, 63, 475 (1990).
- 3. Up to now five human $\alpha(1\rightarrow 3)$ FucTs have been cloned (FucT-III, IV, V, VI and VII), of which the last one is assumed to be participating in the biosynthesis of SLe^x [Nishi *et al. J. Biol. Chem.*, **269**, 14730 (1994)].
- (a) C.-H. Wong; D. P. Dumas, Y. Ichikawa, K. Koseki, S. J. Danishefsky, B. W. Weston and J. B. Lowe, J. Am. Chem. Soc., 114, 7321 (1992); (b) Y.-F. Wang, D. P. Dumas and C.-H. Wong, Tetrahedron Lett., 34, 403 (1993); (c) J. I. Luengo and J. G. Gleason, Tetrahedron Lett., 33, 6911 (1992); (d) S. Cai, M. R. Stroud, S. Hakomori and T. Tokoyuni, J. Org. Chem., 57, 6693 (1992).

$\alpha(1\rightarrow 3)$ FUCOSYLTRANSFERASE

- M. M. Palcic, L. D. Heerze, O. P. Srivastava and O. Hindsgaul, J. Biol. Chem., 264, 17174 (1989).
- 6. P. Smid, G. A. de Ruiter, G. A. van der Marel, F. M. Rombouts and J. H. van Boom, J. Carbohydr. Chem., 10, 833 (1991).
- 7. M. J. Tomaszewski, K. R. Holme and A. S. Perlin, *Carbohydr. Res.* 217, 237 (1991).
- 8. S. Nakabayashi, C. D. Warren and R. W. Jeanloz, *Carbohydr. Res.* 150, C7 (1986).
- G. H. Veeneman, S. H. van Leeuwen and J. H. van Boom, Tetrahedron Lett., 31, 1331 (1990).
- 10. S. Sato, M. Mori, Y. Ito and T. Ogawa, Carbohydr. Res. 155, C6 (1986).
- 11. G. A. van der Marel, Ph.D. Thesis, University of Leiden, 1982.
- 12. N. D. Sinha, J. Biernat, J. McManus and H. Köster, Nucl. Acids Res., 6, 4539 (1984).
- 13. M. H. Caruthers, S. L. Beaucage, J. W. Efkavitch, E. F. Fischer, M. D. Matteuci and Y. Stabinsky, *Nucl. Acids Res. Symp. Ser.*, 7, 215 (1980).
- H. J. G. Broxterman, G. A. van der Marel and J. H. van Boom, *Tetrahedron Lett.*, 29, 4893 (1988).
- M.-J. Camarasa, P. Fernández-Resa, M.-T. Garcia-Lopéz, F. G. De las Heras, P. P. Méndez-Castrillón, B. Alarcón and L. Carrasco, J. Med. Chem., 28, 40 (1985).